tence should read but which may originate from addition of the aminocyanocarbene **11** to the tautomeric aminocyanoketenimine form of **1**.

Page 4360. In the text, column 2, the fourth paragraph now reads "The enaminoimine **23** may originate from insertion of the carbene **11** into the C–H bond of **1**, as proposed earlier[1] for formation of **23** from **1** in triethylamine." This paragraph should read The enaminoimine **23** may originate from addition of the carbene **11** to **2**—the tautomeric aminocyanoketenimine form of **1**—as proposed earlier for formation of **23** from **1** in triethylamine.

Y. Sato,* Y. Ban, and H. Shira: Synthesis of *N*-(2-Triphenylstannylethyl)amines and Their Reactivities.

Page 4376. In Table III, sixth entry, replace "Bz" by $C_6H_5CH_2$ in second and fifth columns

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M. J. Robins* and R. A. Jones: Nucleic Acid Related Compounds. 9. The Synthesis of 6-Amino-9-(2-deoxy-D-*erythro*-pent-1-enofuranosyl)purine, the First 1′,2′-Unsaturated Purine Nucleoside.

Page 115. Column 1, paragraph 2, line 4. "9-(5-methyl-2-furyl)adenine[3a]" should read 9-(5-hydroxymethyl-2-furyl)adenine.

W. E. Truce,* D. L. Heuring, and G. C. Wolf: Addition of Sulfonyl Iodides to Allenes.

Page 241. Column 1, Scheme II. The one pair of bracketed formulas in scheme II should be interchanged with the other pair.

George L. Hardgrove, Jr.,* J. Stuart Bratholdt, and Mary M. Lien: The Crystal Structure of *cis*-2,4-Diphenylthietane *trans*-1-Monoxide.

Page 246. Line 3. Third author's name should read Lien (not Lein).

Kennth L. Marsi: Phenylsilane Reduction of Phosphine Oxides with Complete Stereospecificity.

Page 267. Column 1, line 28. The compound used was (−)-(*S*)p-menthyl methylphenylphosphinate.

Page 267. Column 1. Reference 2b should be H. Christol and H.-J. Cristau, *Ann. Chim.*, **6,** 191 (1971).

Donald J. Hart and Warren T. Ford*: Carbon-13 Nuclear Magnetic Resonance Spectra of Tetraalkylammonium Tetraalkylborides.

Page 365. Column 2, Table IV. Below the title and above the data in the table should be the equation $\delta_C{}^X = A(\delta_C{}^{\mathrm{alkane}}) + B$.

Joseph San Filippo, Jr,* and G. M. Anderson: The Reduction of 2-Substituted 2-Halonorbornanes by Tri-*n*-butyltin Hydride.

Page 474. Table I. Footnote g should be as follows. [g] Taken from ref 38.

M. P. Cava,* H. Firouzabadi, and M. Krieger: 1,2-Diphenylanthra[*b*]cyclobutadiene.

Page 480. In structure **13**, the lower four-membered ring is incorrectly drawn. It should be the same as the upper four-membered ring, *i.e.*, an exomethylenecyclobutenone.

N. C. Deno,* K. A. Eisenhardt, D. G. Pohl, H. J. Spinnelli, and R. C. White: Photochlorination of Alcohols.

Page 520. We wish to call specific attention here to the work of Dr. Kollonitsch and coworkers, comprising the first synthesis of chloro alcohols from alcohols by free-radical chlorination in acid media [J. Kollonitsch, G. A. Doldouras, and V. F. Verdi, *J. Chem. Soc., B*, 1093 (1967), ref 11 in our paper].

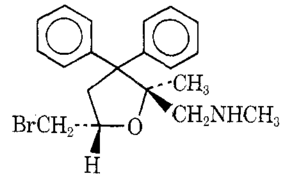
I. Angres and H. E. Zieger: The Reaction of Lithium Naphthalenide with Quaternary Ammonium Salts.

Page 1013. The premise " . . . the mole percentage of unreacted salt (Table I) is a measure of the maximum amount of carbanion formed by path B . . ." cannot possibly be correct. It may be a measure of the amount of carbanion that escaped dimer formation and appeared as % RH in Table I. Consequently, the conclusion that *dimer forms by both path A and path B is not supported by the* data. The only evidence available indicates path B, and path A remains an undemonstrated possibility.

The % $(CH_3)_3N$ for compound **3** in Table I is 81 not 18.

Henry Wong, Jacques Chapuis, and Ivo Monkovic*: Stereochemical Course of Bromocyclizations of γ,δ-Unsaturated Alcohols. II. Approaches to Various Oxaazabicyclooctane and -nonane Systems.

Page 1043. The structure given for **20** is erroneous; the correct structure is shown below.



20

R. B. Nelson and G. W. Gribble*: Reduction of Aryl Iodides with Sodium Hydride.

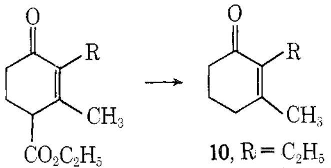
Page 1425. Column 3, Table I. The yield for conversion of *o*-iodobenzoic acid to benzoic acid should read 95%.

Dwain M. White: 2,5-Dicarbomethoxy-3,4-diphenylcyclopentadienone. Synthesis and Reaction with Acetylenes.

Page 1951. Reference should be made to the synthesis and characterization of the title compound and its precursor 2,5-dicarbo*methoxy-4-hydroxy-3,4-diphenylcyclopenten-2-one* by R. C. Cookson, J. B. Henstock, J. Hudec, and B. R. D. Whitear, *J. Chem. Soc.* 1989 (1967), and B. Eistert and A. J. Thommen, *Chem. Ber.*, **104,** 3048 (1971).

Edward J. Parish, Naresh V. Mody, Paul A. Hedin, and D. Howard Miles*: Cleavage of δ-Keto β,α-Unsaturated Esters by 1,4-Diazabicyclo[2.2.2]octane.

Page 1593. Column 1, Scheme I. For structures **9–12** the R group was not located. Structures **9–12** should be as follows.

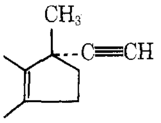


9, R = C_2H_5
11, R = H

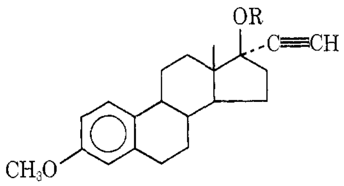
10, R = C_2H_5
12, R = H

Ramesh M. Kanojia,* Linda Yarmchuck, and Irving Scheer: Epimerization of Mestrand Acetate on Alumina.

Page 2304. Column 2. The formulas for the structures **10** and **1** are in error showing an extra methyl group at C_{13} in **10** and at C_{10} in **1** and should be correctly represented as follows.



10



1

C. Wong and W. W. Paudler*: Synthesis and Conformation of [2.2](2,5)Furano(2,5)pyridinophane.

Page 2570. In the abstract, "C_3–C_4" should be N–C_6, and "N–C_6" should be C_3–C_4.

Page 2570. Column 2, Table I. The first entry should be **2** (opposite 48%); the second entry should be **1a** instead of "**21a**" (opposite 18%).

R. A. Hites: Phytadienes from the Pyrolysis of Pheophytin a.

Page 2634. In our paper we indicated "It is possible that the phytadienes reported to be present in zooplankton (Blumer and Thomas, *Science*, 1965) may have been an artifact." Dr. Blumer has provided the following additional information to indicate that the zooplankton phytadienes are not artifacts: "They were isolated and resolved after cold pentane extraction from zooplankton by chromatography over deactivated silica gel, under conditions demonstrated neither to dehydrate phytol to phytadienes nor to isomerize the olefins. Thus, the proof of their presence does not hinge on exposure to gc temperatures. Where gc was used to determine amounts or to trap the olefins, it was applied to chromatographed fractions that contained neither phytol nor pheophytin or other chlorophyll derivatives. Injector temperature was at or below 200° and it was demonstrated that isomerization of the olefins did not take place."